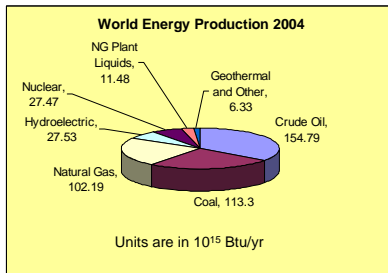


Energy Balances on Complex Fuels

- Why worry about complex fuels?
 - Crude oil, coal, wood, biomass, etc.



Hypothetical Coal Molecule

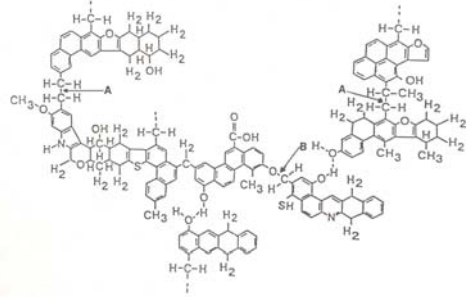


Figure 1. Summary of coal structure information in a hypothetical coal molecule.
From Solomon and coworkers (1985)

Reacting Coal Molecule

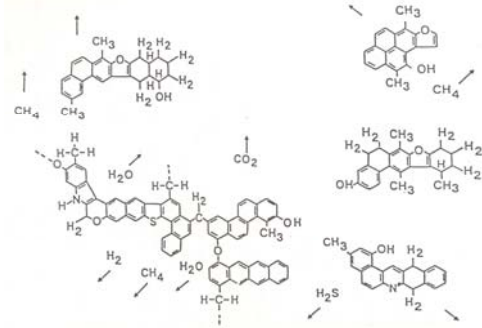


Figure 2. Cracking of hypothetical coal molecule during thermal decomposition.

Fuel Characterization

Where are...graphite...oil...peat...biomass?

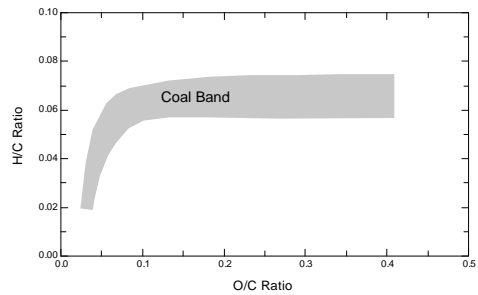


TABLE I. Classification of Coals by Rank

Class	Group	Fixed carbon limits (%) (dry, mineral-matter-free basis)		Volatile matter limits (%) (dry, mineral-matter-free basis)		Calorific value limits (Btu/lb) (moist, mineral-matter-free basis)		Agglomerating character
		≥	<	≥	<	≥	<	
I. Anthracite	1. Meta-anthracite	98	—	—	2	—	—	nonagglomerating
	2. Anthracite	92	98	2	8	—	—	
	3. Semianthracite	86	92	8	14	—	—	
II. Bituminous	1. Low volatile bituminous coal	78	86	14	22	—	—	commonly agglomerating
	2. Medium volatile bituminous coal	69	78	22	31	—	—	
	3. High volatile A bituminous coal	—	69	31	—	14,000	14,000	
	4. High volatile B bituminous coal	—	—	—	—	11,500	13,000	
	5. High volatile C bituminous coal	—	—	—	—	10,500	11,500	
III. Subbituminous	1. Subbituminous A coal	—	—	—	—	10,500	11,500	nonagglomerating
	2. Subbituminous B coal	—	—	—	—	9,500	10,500	
	3. Subbituminous C coal	—	—	—	—	8,300	9,500	
IV. Lignite	1. Lignite A	—	—	—	—	6,300	8,300	nonagglomerating
	2. Lignite B	—	—	—	—	—	6,300	

Terminology

- High heating value
 - Calculated using H_2O (liq) as product
- Low heating value
 - Calculated using H_2O (gas) as product
- Heating value = $-\Delta H_c$
 - i.e., heating value is positive, but heat of reaction is negative