

## Class 4

### Transient & Steady-State Balances

1

## HW Mistakes

- Finding the weight in lbf from lbm. Many students forgot to convert to lb<sub>f</sub> using  $32.2 \frac{\text{lb}_m \text{ft}}{\text{lb}_f \text{s}^2}$  after calculating  $m \cdot g$
- Not recognizing that  $A \cdot \exp(a) = A \cdot e^a$ 
  - Many students incorrectly interpreted  $A \cdot \exp(a)$  as  $A^a$

2

## Math Refresher

### Definite integrals

We often use **definite integrals** rather than expand with constants and set boundary conditions. The beginning values are at the start of the integral and the end values are at the end of the integral.

$$\int_3^5 x dx = \left. \frac{x^2}{2} \right|_3^5 = \frac{5^2}{2} - \frac{3^2}{2} = \frac{(25 - 9)}{2} = \frac{16}{2} = 8$$

Remember that

$$\ln a - \ln b = \ln \frac{a}{b},$$

and

$$\int \frac{1}{(a+bx)} dx = \frac{1}{b} \ln(a+bx),$$

so

$$\int_{x_1}^{x_2} \frac{1}{(a+bx)} dx = \frac{1}{b} [\ln(a+bx_2) - \ln(a+bx_1)] = \frac{1}{b} \ln \left( \frac{a+bx_2}{a+bx_1} \right)$$

3

## Classification of Balances

- Continuous
- Batch
- Semibatch
- Steady-state
- Transient

Give me an example of each...

4

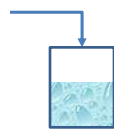
## General Balance Equation

$$\text{Accumulation} = \text{In} - \text{Out} + \text{Generation} - \text{Consumption}$$

How do you remember this equation?

5

## Take terms 2 at a time



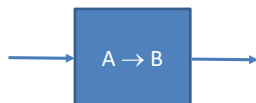
Semibatch  
Transient  
 $\text{In} = \text{Accumulation}$



Semibatch  
Transient  
 $\text{Accumulation} = -\text{Out}$



Continuous  
Steady-state  
 $\text{In} = \text{Out}$



Continuous  
Steady-state  
 $\text{In}_A = \text{Consumption}_A$   
 $\text{Generation}_B = \text{Out}_B$  (100% Conversion of A)

6

## Cross Out Terms That Are Zero

### Steady State

$$\text{Accumulation} = \text{In} - \text{Out} + \text{Generation} - \text{Consumption}$$

At steady state, all time-dependent terms are zero

### Mass Balance

$$\text{Accumulation} = \text{In} - \text{Out} + \text{Generation} - \text{Consumption}$$

You cannot generate or destroy mass unless you have a nuclear reaction

### Mole Balance

$$\text{Accumulation} = \text{In} - \text{Out} + \text{Generation} - \text{Consumption}$$

Generation and consumption relate to chemical reaction terms

7

### Example: Donut Eatery



In

Consumption

Out

So:

$$\text{In} = \text{Out} + \text{Consumption}$$

8

## Transient Balances

$$accum = in - out + gen - cons$$

- If we sum over time, then the change in the system has to equal the total amount that came in minus the amount that went out, adjusted for generation and consumption
- If we pour a glass of milk, then the “accumulation” is the amount that ended up in the glass
- The “in” term is the amount that we poured into the glass
- The “out” term might be the amount sipped through a straw

If we are interested in the time-dependent behavior, we want the amount of change per time

Change in our variable  $y$  per time is  $\frac{\Delta y}{\Delta t}$  which in the limit becomes  $\frac{dy}{dt}$

9

## Transient Balances

$$accum = in - out + gen - cons$$

The accumulation term always involves a derivative:  $accum = \frac{d(?)}{dt}$  or  $\frac{\Delta y}{\Delta t}$

What are the units of the denominator of the derivative term  $\frac{d}{dt}$ ? dt has units of time

**Example:** total mass

$$\frac{dm}{dt} = \dot{m}_{in} - \dot{m}_{out}$$

**Other relationships:**

$$m = \rho V$$

$$mass = \frac{mass}{volume} \times volume$$

$$\dot{m} = \rho v A = \rho \dot{V} \quad \frac{mass}{time} = \frac{mass}{volume} \times \frac{volume}{time}$$

10

## Separate and Integrate

### (Solve the ODE)

- **A. Separate** means put one variable on left side and the other variable on the right side.
- Example:  $\frac{dy}{dt} = f(y) \cdot f(t)$
- Split the  $dy$  and the  $dt$  (like a quotient), and then put  $dt$  on the RHS.
- Also put all terms with a  $y$  on the LHS.

$$\frac{dy}{f(y)} = f(t) \cdot dt$$

More info in Chapter 10 (10.1-10.2)

11

## Integrate

- **B. Integrate** both sides between limits that correspond to each other.

$$\int_{y_0}^{y_f} \frac{dy}{f(y)} = \int_{t_0}^{t_f} f(t) dt$$

Boundary conditions:

$$y = y_0 \text{ at } t = t_0$$

$$y = y_f \text{ at } t = t_f$$

- Example:  $\frac{dy}{dt} = -yt^2$

- Separate:  $\frac{dy}{y} = -t^2 dt$

- Integrate:  $\int_{y_0}^{y_f} \frac{dy}{y} = - \int_{t_0}^{t_f} t^2 dt$

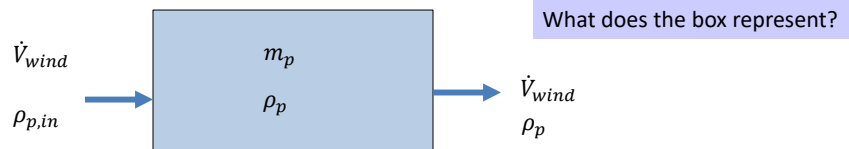
- or  $\ln\left(\frac{y_f}{y_0}\right) = -\frac{1}{3}(t_f^3 - t_0^3)$   
 $y_f = y_0 e^{-\frac{1}{3}(t_f^3 - t_0^3)}$

Note that I used definite integrals here.  
This is MUCH easier and faster.  
Please learn how to do this!!!!

12

## Example: Air Quality in Utah Valley

- Let  $m_p$  = mass of pollutants (kg of pollutants)
- Let  $\rho_p$  = mass concentration of pollutants (kg of p/m<sup>3</sup>)
- Assume valley is a vessel
- Wind comes along
- Perfectly mixed ( $\rho_p = \rho_{p,out}$ )
- Find  $\rho_p$  as a function of time



13

## Species Balance on “p”

Accumulation = In – Out + Generation - Consumption

$$\frac{dm_p}{dt} = \dot{m}_{p,in} - \dot{m}_{p,out} \quad = \frac{\Delta m_p}{\Delta t}$$

$$m_p = \rho_p V_{\text{valley}}$$

$$\dot{m}_p = \rho_p vA = \rho_p \dot{V}$$

So the balance becomes:

$$\frac{d\rho_p V}{dt} = \rho_{p,in} \dot{V}_{wind} - \rho_p \dot{V}_{wind} = \dot{V}_w (0 - \rho_p)$$

Assume  $\rho_{p,in} = 0$

$$V \frac{d\rho_p}{dt} = -\rho_p \dot{V}_w$$

14

## Now Separate!

$$V \frac{d\rho_p}{dt} = -\rho_p \dot{V}_w$$



$$\frac{d\rho_p}{\rho_p} = -\frac{\dot{V}_w}{V} dt$$

15

## Now Integrate

$$\int_{\rho_{p,0}}^{\rho_{p,f}} \frac{d\rho_p}{\rho_p} = -\frac{\dot{V}_w}{V} \int_0^t dt$$

$$\ln\left(\frac{\rho_{p,f}}{\rho_{p,0}}\right) = -\frac{\dot{V}_w}{V} t$$

$$\rho_{p,f} = \rho_{p,0} e^{-\frac{\dot{V}_w}{V} t}$$

16



## Consistency of Units

- In any equation, each term must have the same units (as we mentioned before)

- Consider the species mass balance equation:

$$\frac{d(?)}{dt} = \dot{m}_{a,in} - \dot{m}_{a,out}$$

- What should be the units where the “?” occurs?

Ans: mass of component “a” ( $m_a$ ), NOT total mass of “a” per time ( $\dot{m}_a$ )

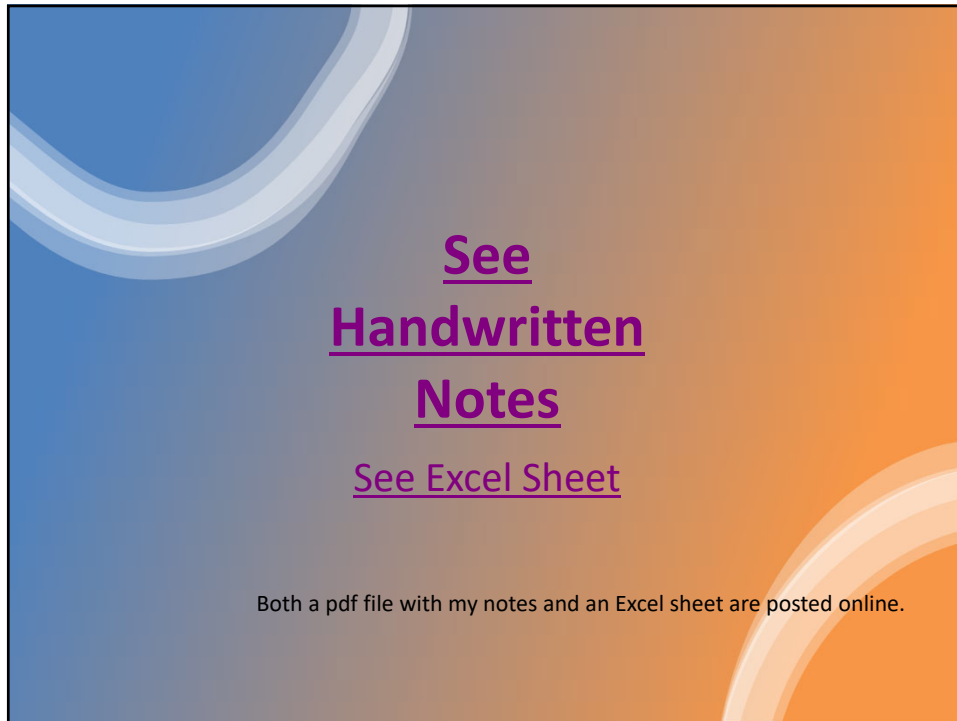
- Getting the units right will help in determining what goes inside the derivative term
  - Only possible terms inside derivative are:  $m$ ,  $m_i$ ,  $n$ ,  $n_i$

17

### Example: Leaking Reactor

- A laboratory reactor leaks  $\text{SO}_2$  at a rate of 0.1 mol/hr. The room volume is 1000  $\text{m}^3$ . If the air/ $\text{SO}_2$  gas leaves the room at the same rate as the leak (0.1 mol/hr), how long will it take for the room to reach the toxic limit of 1 ppm ( $y_{\text{SO}_2} = 1 \times 10^{-6}$ )?
- If the room is ventilated at 10 room volumes per hour, how will  $y_{\text{SO}_2}$  change with time?

18



**See**  
**Handwritten**  
**Notes**  
**See Excel Sheet**

Both a pdf file with my notes and an Excel sheet are posted online.

19

## Integral Balances

- Commonly used on batch processes
- Balance initial and final states
- No time dependence
- No in and out streams

80% A
20% B
0% C



40% A
40% B
20% C

### Balance on A

$$0.8 \text{ m} = 0.4 \text{ m} + \text{consumption}$$

Or

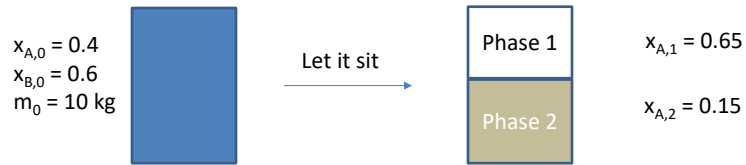
$$\text{Consumption} = 0.8\text{m} - 0.4\text{m} = 0.4 \text{ m}$$

**Initial input + generation = final output + consumption**

Note: Total mass is constant in a batch process

20

## Example of Integral Balances



Find masses of phase 1 and phase 2

- Total mass balance

$$m_0 = 10 \text{ kg} = m_1 + m_2$$

- Balance on species A

$$x_{A,0} m_0 = x_{A1} m_1 + x_{A2} m_2$$

or

$$0.4 (10 \text{ kg}) = 0.65 m_1 + 0.15 m_2$$

21

## Algebra

$$10 = m_1 + m_2$$

$$4 = 0.65 m_1 + 0.15 m_2$$

$$1.5 = 0.15 m_1 + 0.15 m_2$$

$$2.5 = 0.5 m_1$$

Multiply by 0.15

Now subtract to eliminate  $m_2$

$$\text{So } m_1 = 2.5/0.5 = 5 \text{ kg,}$$

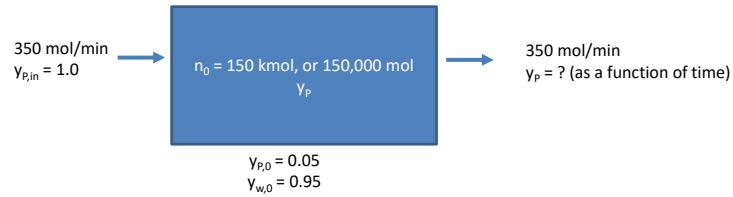
$$\text{and therefore } m_2 = 10 - 5 = 5 \text{ kg}$$

22



Click to see problem

## HW 5.3



$$\text{Accum} = \text{in} - \text{out} + \text{gen} - \text{cons}$$