

## Crazy Test Answers

### HOW NOT TO ANSWER TEST QUESTIONS

<p>Are you childish?</p> <p><input type="checkbox"/> Yes</p> <p><input checked="" type="checkbox"/> No</p>	<p>Are you drunk?</p> <p><input type="checkbox"/> Yes</p> <p><input checked="" type="checkbox"/> No</p>
<p>Are you paranoid?</p> <p><input type="checkbox"/> Yes</p> <p><input checked="" type="checkbox"/> No, WHY?</p>	<p>Are you stupid?</p> <p><input type="checkbox"/> Yes</p> <p><input checked="" type="checkbox"/> No</p>

1

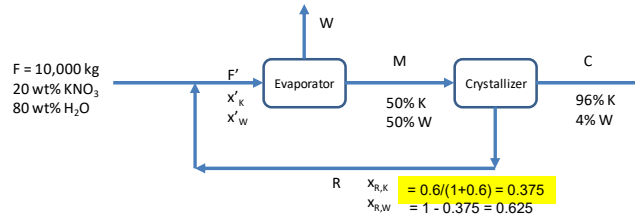
## Class 9 Practice Multi-Unit DOF

- Questions?
- 3 Worksheets
- Problem 4.31



2

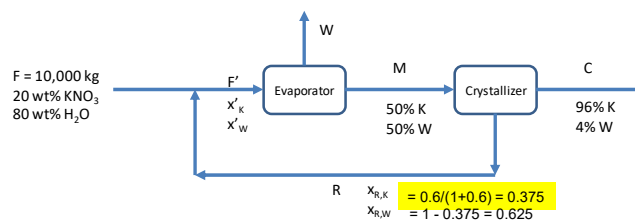
1. Fresh feed containing 20% by weight  $\text{KNO}_3$  in  $\text{H}_2\text{O}$  is combined with a recycle stream and fed to an evaporator where the water content is reduced. The concentrated solution leaving the evaporator, containing 50%  $\text{KNO}_3$ , is fed to a crystallizer. The crystals obtained from the crystallizer are 96%  $\text{KNO}_3$  and 4% water. The supernatant liquid (liquid in equilibrium with crystals) from the crystallizer constitutes the recycle stream and contains 0.6 kg  $\text{KNO}_3$  per 1.0 kg of  $\text{H}_2\text{O}$ . The objective is to complete the mass balance showing all stream values and compositions.



	Unknowns	# of U	BE	OE	DOF
Evap					
Cryst					
Mixer					
Overall					

3

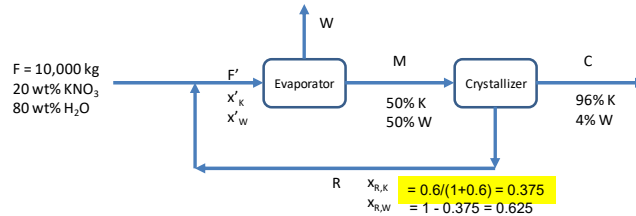
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	Unknowns	# of U	BE	OE	DOF
Evap	$F', x'_K, W, M$	4	2	0	2
Cryst					
Mixer					
Overall					

4

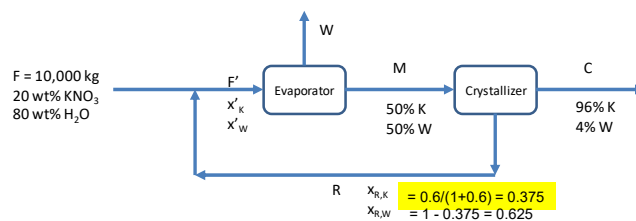
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	Unknowns	# of U	BE	OE	DOF
Evap	$F', x'_K, W, M$	4	2	0	2
Cryst	$M, R, C$	3	2	0	1
Mixer					
Overall					

5

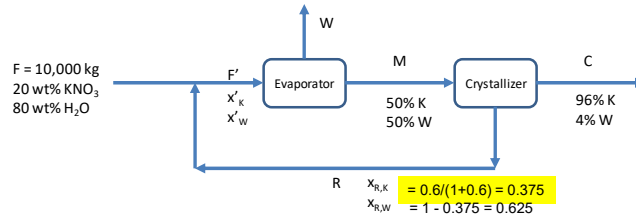
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	Unknowns	# of U	BE	OE	DOF
Evap	$F', x'_K, W, M$	4	2	0	2
Cryst	$M, R, C$	3	2	0	1
Mixer	$R, x'_K, F'$	3	2	0	1
Overall					

6

1. Fresh feed containing 20% by weight  $\text{KNO}_3$  in  $\text{H}_2\text{O}$  is combined with a recycle stream and fed to an evaporator where the water content is reduced. The concentrated solution leaving the evaporator, containing 50%  $\text{KNO}_3$ , is fed to a crystallizer. The crystals obtained from the crystallizer are 96%  $\text{KNO}_3$  and 4% water. The supernatant liquid (liquid in equilibrium with crystals) from the crystallizer constitutes the recycle stream and contains 0.6 kg  $\text{KNO}_3$  per 1.0 kg of  $\text{H}_2\text{O}$ . The objective is to complete the mass balance showing all stream values and compositions.

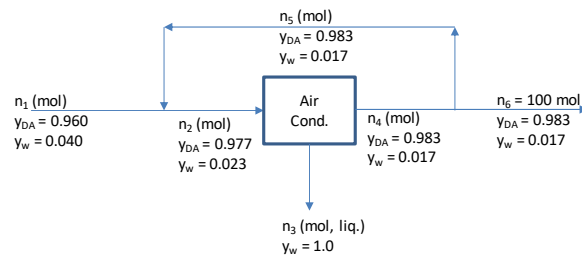


	Unknowns	# of U	BE	OE	DOF
Evap	$F', x'_K, W, M$	4	2	0	2
Cryst	$M, R, C$	3	2	0	1
Mixer	$R, x'_K, F'$	3	2	0	1
Overall	$W, C$	2	2	0	0

How do you start?

7

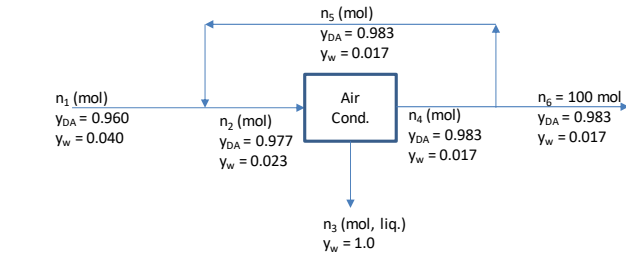
2.



	Unknowns	# of U	BE	OE	DOF
Air cond					
Splitter					
Mixer					
Overall					

8

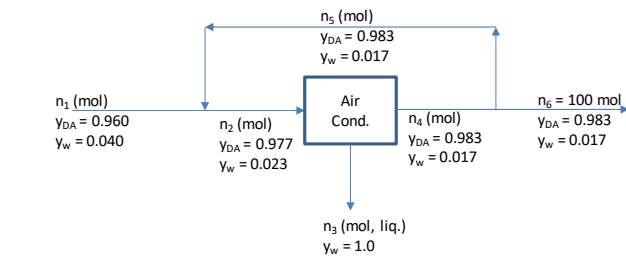
2.



	Unknowns	# of U	BE	OE	DOF
Air cond	$n_2, n_3, n_4$	3	2	0	1
Splitter					
Mixer					
Overall					

9

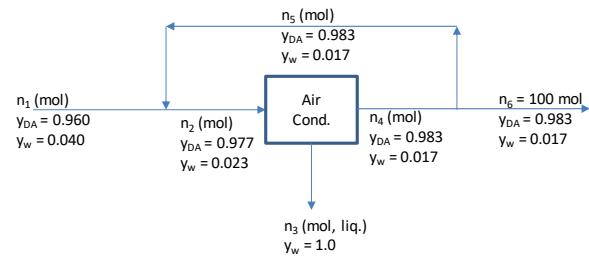
2.



	Unknowns	# of U	BE	OE	DOF
Air cond	$n_2, n_3, n_4$	3	2	0	1
Splitter	$n_4, n_5$	2	1	0	1
Mixer					
Overall					

10

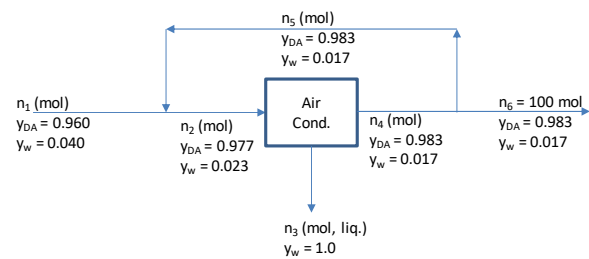
2.



	Unknowns	# of U	BE	OE	DOF
Air cond	$n_2, n_3, n_4$	3	2	0	1
Splitter	$n_4, n_5$	2	1	0	1
Mixer	$n_1, n_2, n_5$	3	2	0	1
Overall					

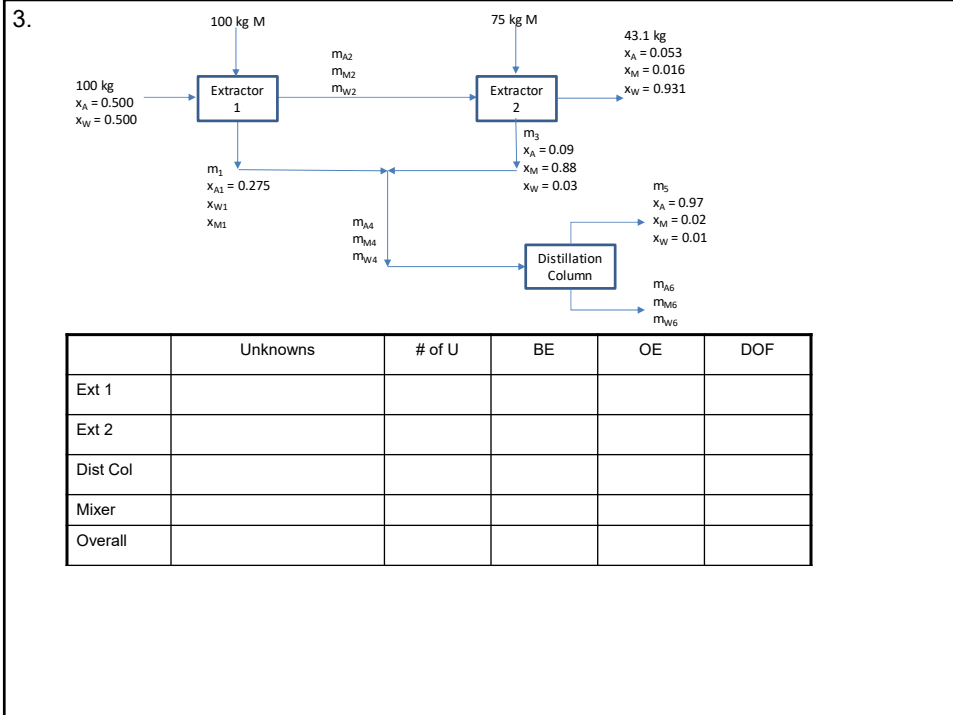
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2.

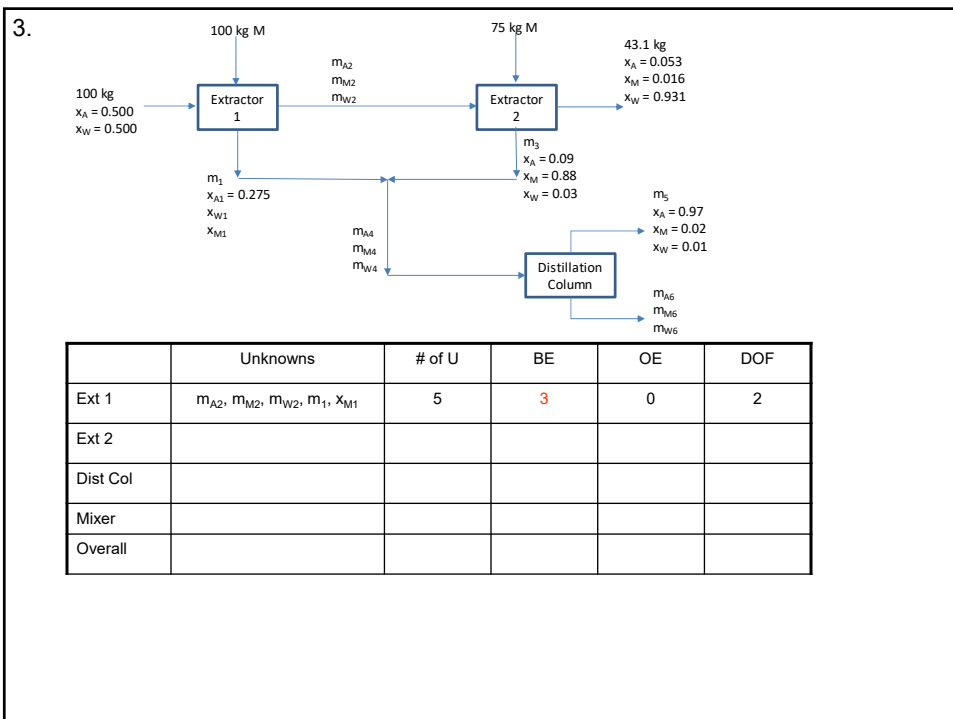


	Unknowns	# of U	BE	OE	DOF
Air cond	$n_2, n_3, n_4$	3	2	0	1
Splitter	$n_4, n_5$	2	1	0	1
Mixer	$n_1, n_2, n_5$	3	2	0	1
Overall	$n_1, n_3$	2	2	0	0

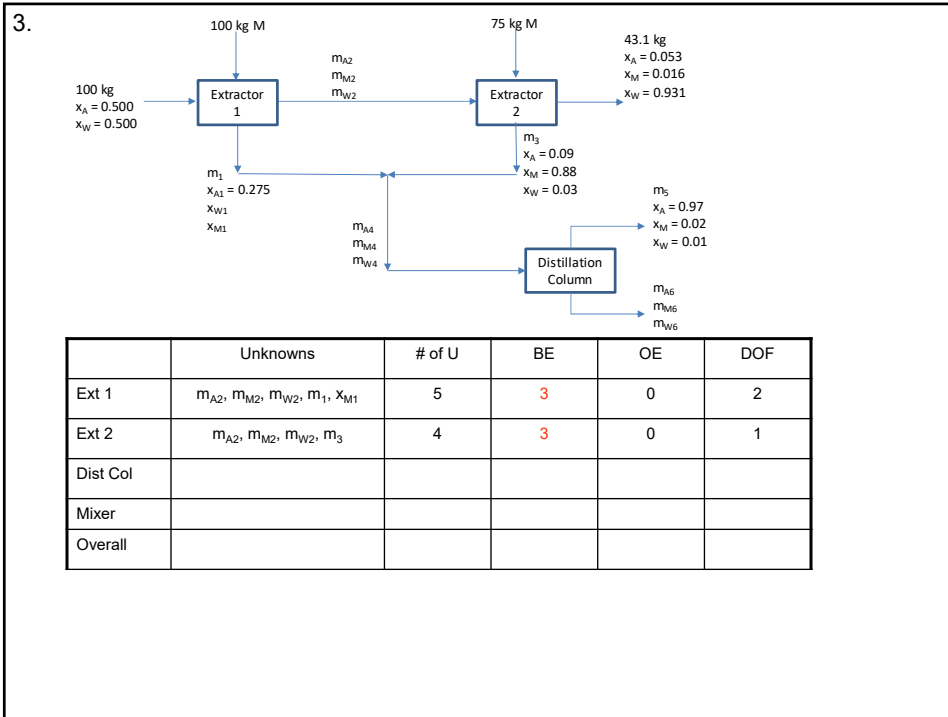
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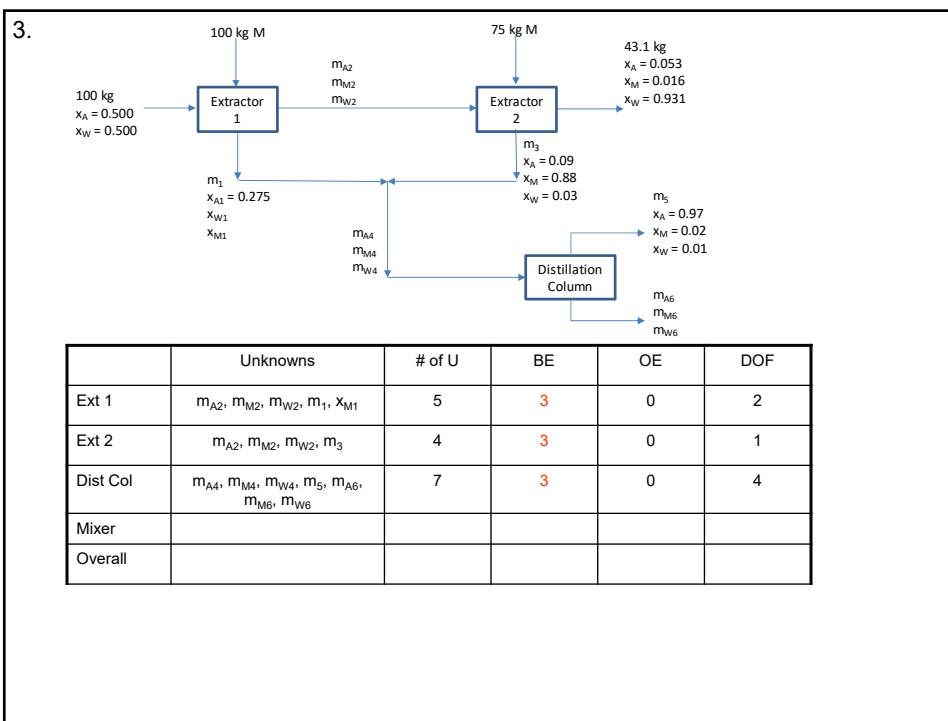
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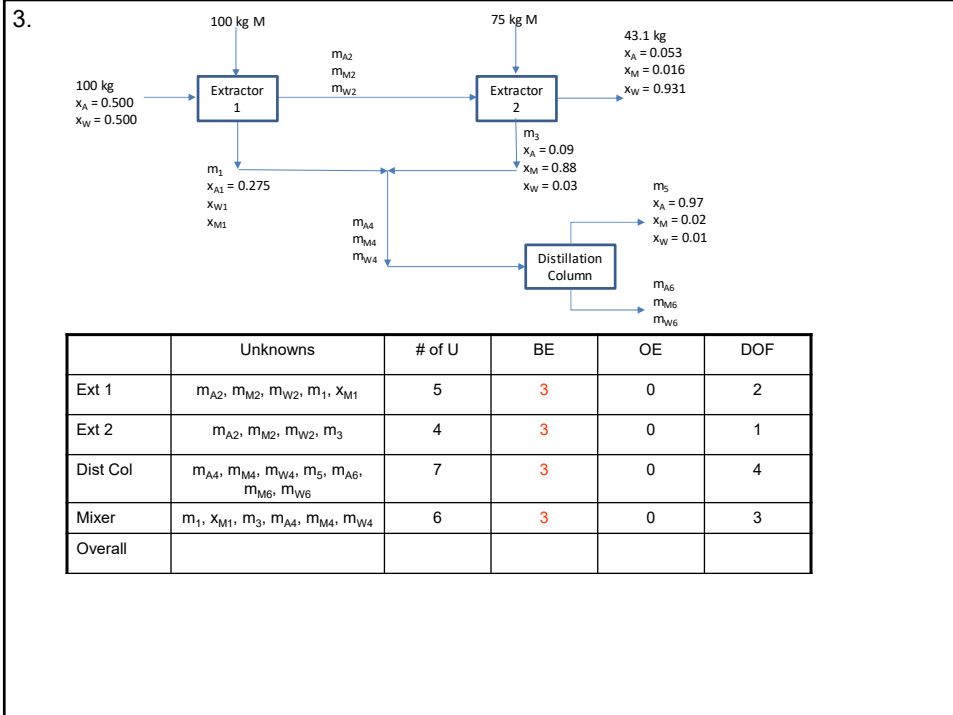


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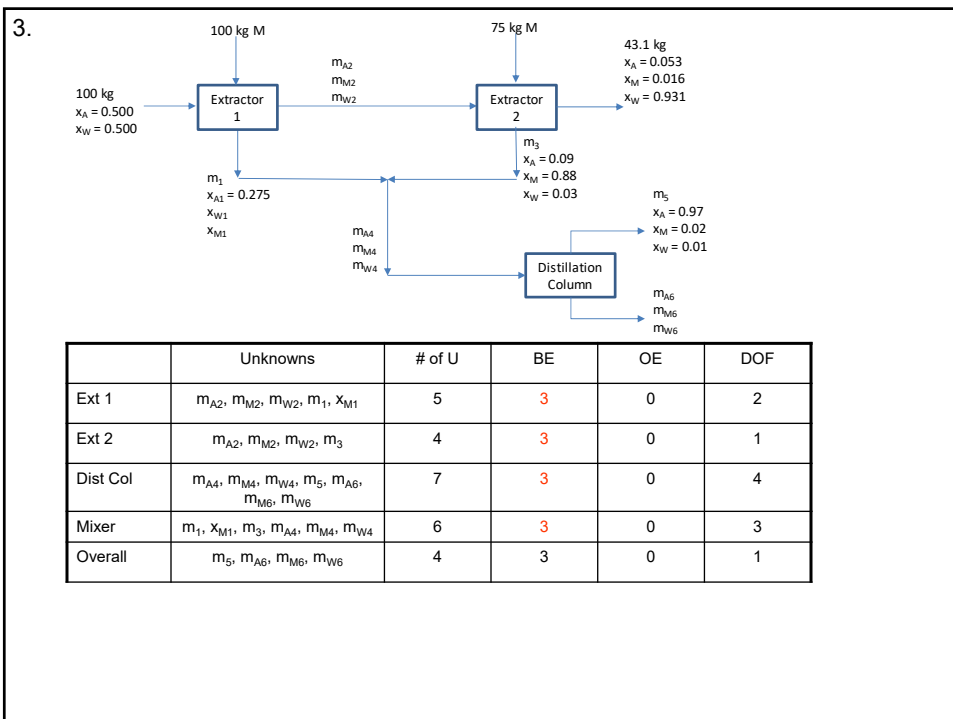


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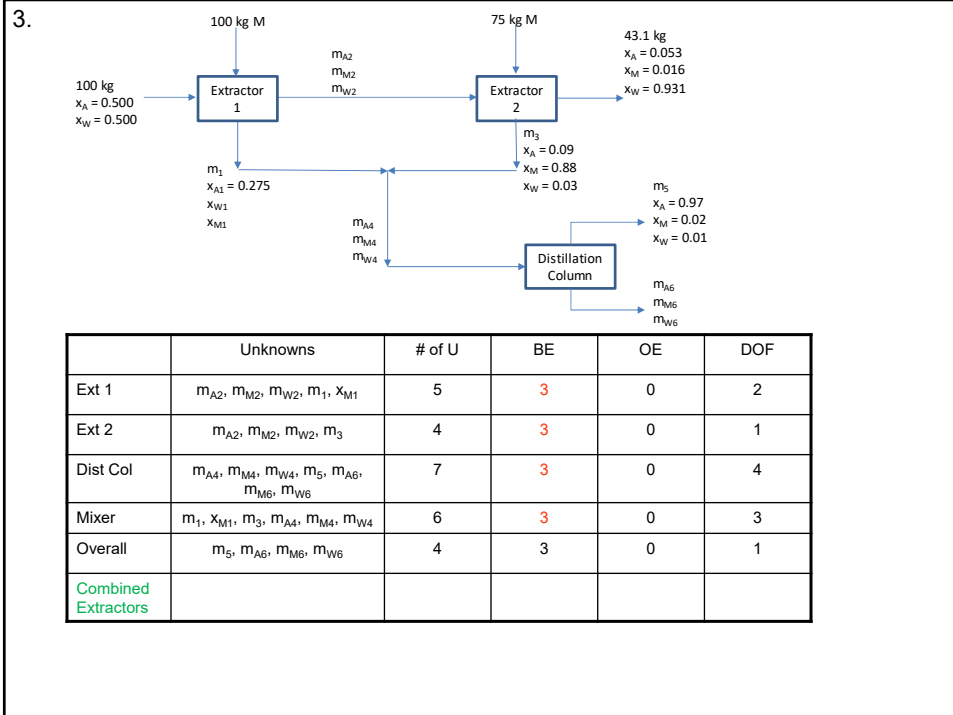




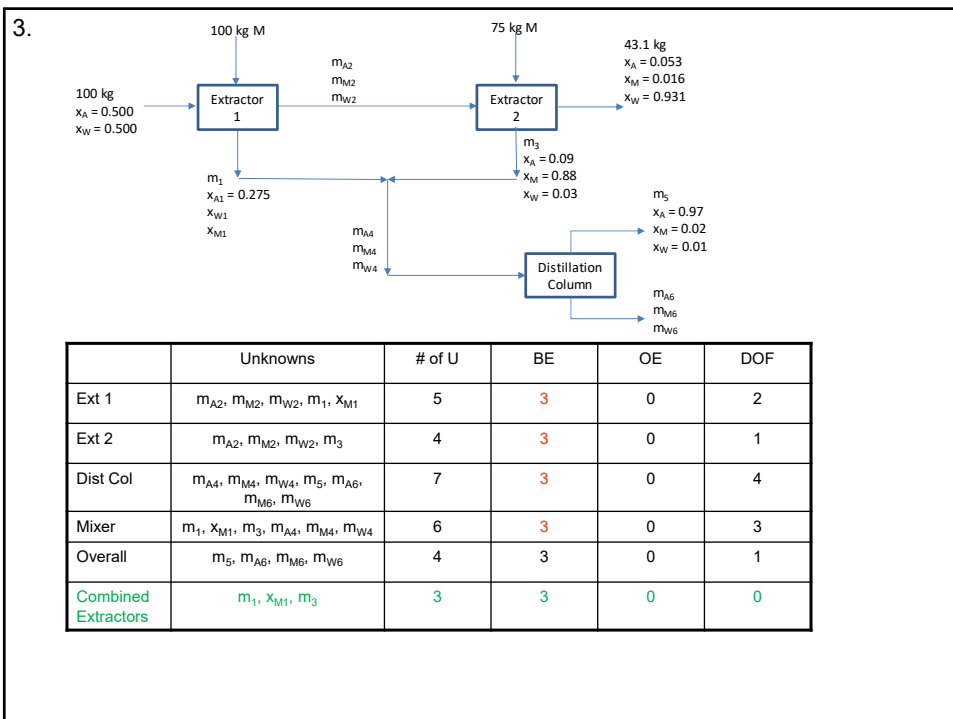
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18



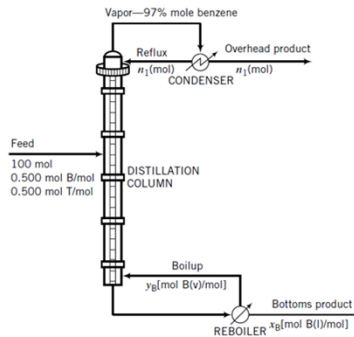
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9.1. An equimolar liquid mixture of benzene and toluene is separated into two product streams by distillation. A process flowchart and a somewhat oversimplified description of what happens in the process follow:

4.31 (3<sup>rd</sup> Ed)  
4.42 (4<sup>th</sup> Ed)



Inside the column a liquid stream flows downward and a vapor stream rises. At each point in the column some of the liquid vaporizes and some of the vapor condenses. The vapor leaving the top of the column, which contains 97 mole% benzene, is completely condensed and split into two equal fractions: one is taken off as the overhead product stream, and the other (the **reflux**) is recycled to the top of the column. The overhead product stream contains 89.2% of the benzene fed to the column. The liquid leaving the bottom of the column is fed to a partial reboiler in which 45% of it is vaporized. The vapor generated in the reboiler (the **boilup**) is recycled to become the rising vapor stream in the column, and the residual reboiler liquid is taken off as the bottom product stream.

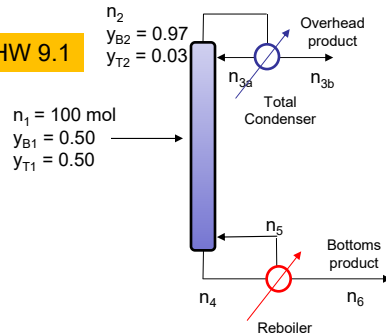
The compositions of the streams leaving the reboiler are governed by the relation:

$$\frac{y_B/(1-y_B)}{x_B/(1-x_B)} = 2.25$$

where  $y_B$  and  $x_B$  are the mole fractions of benzene in the vapor and liquid streams, respectively.

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### HW 9.1



A. Vapor leaving the top is completely condensed and split into two equal fractions

B. Overhead product stream contains 89.2% of benzene fed to the column.

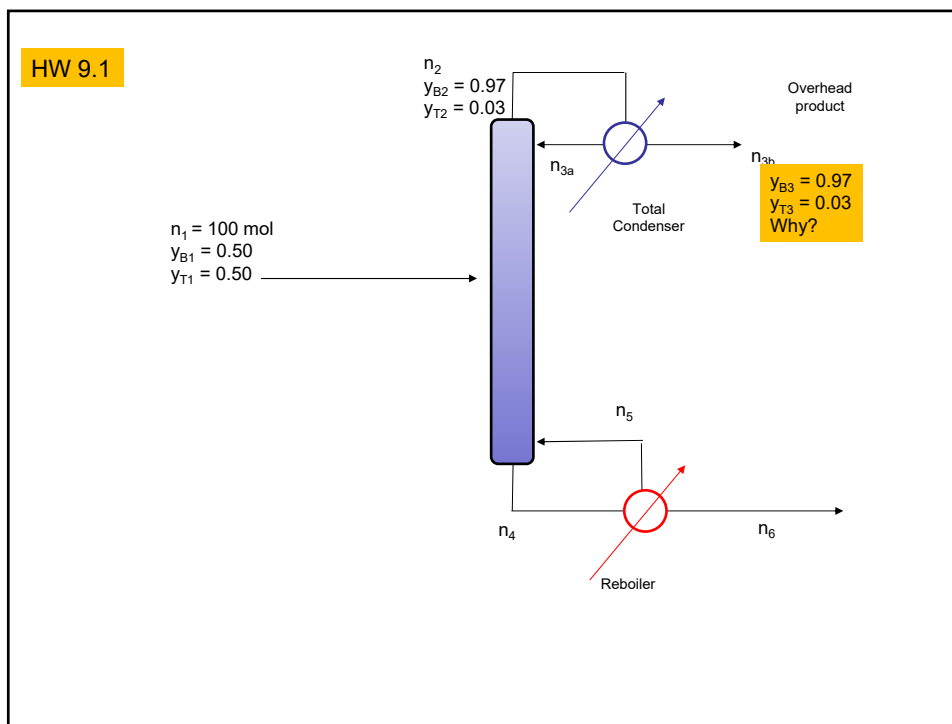
C. Liquid leaving the bottom of the column is fed to a partial reboiler in which 45% of it is vaporized. ("It" means liquid stream fed to reboiler, or  $n_4$ )

D. The composition of the streams leaving the reboiler are governed by the equation:

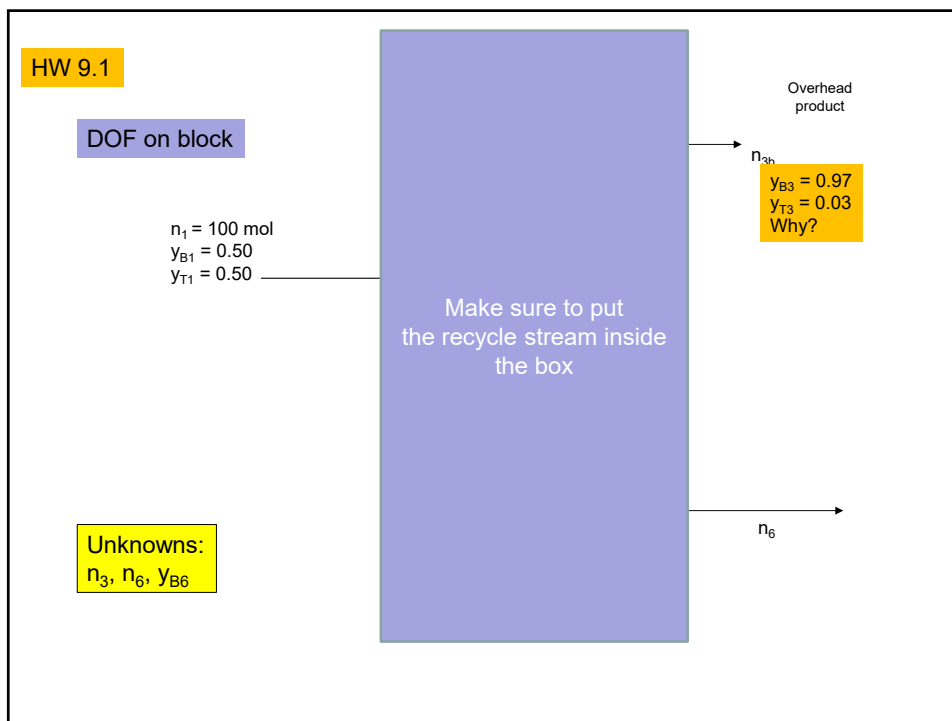
$$\frac{y_B/(1-y_B)}{x_B/(1-x_B)} = 2.25$$

1. Which streams are gas and which are liquid?
2. What are the different subunits for balance equations?
3. What are the unknowns in each subunit?
4. Translate the clues into "additional equations".
5. Perform the DOF analysis on each unit.
6. Write out the logic (with equations) that will solve for all unknown variables.
7. Solve for the 4 variables asked for in the problem statement (what variables are wanted?).

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