



Concepts

1. Total energy is conserved
2. Energy can transform from one form to another

Forms

1. Kinetic $E_k = \frac{1}{2}mv^2$
 2. Potential $E_p = mgh$
 3. Internal energy $U = f(T)$
~~may~~ may also be $f(P, V)$
- } force x distance

Closed Systems

no mass crosses the boundary

1. Heat (energy flow caused by ΔT)

Q = positive when heat is transferred to the system from surroundings

2. Work

W = positive when work is done by the system on the surroundings

Types

PV (mechanical)

friction

shaft

flow

electrical

open

$$\text{in} + \cancel{\text{gen}} = \text{out} + \text{accum} + \cancel{\text{cons}}$$

$$\Delta U + \Delta E_k + \Delta E_p = Q - W$$

SI: J or kJ

cgs ergs

kcal 1 kg H₂O 1°C

American: Btu 1 Btu = 1 lbm H₂O(l) 1°F

Correct the ppt

7.5 (7.6 in 4th Ed)

Air at 300°C and 130 kPa flows through a horizontal 7-cm ID pipe at a velocity of 42.0 m/s.

- Calculate $\dot{E}_k$ in Watts, assuming ideal gas behavior.
- If the air is heated to 400°C at constant pressure, what is $\Delta\dot{E}_k (= \dot{E}_k(400^\circ\text{C}) - \dot{E}_k(300^\circ\text{C}))$?
- Why would it be incorrect to say that the rate of transfer of heat to the gas in part (b) must equal the rate of change of kinetic energy?

Air $\rightarrow$
300°C
130 kPa

$d = 7\text{ cm}$
 $V = 42\text{ m/s}$

$\dot{E}_k = \frac{1}{2} \dot{m} v^2$
 $\dot{m} = \rho A v$

$$\rho_{\text{air}} = \frac{P M W}{R T} = \frac{(130 \times 10^3 \text{ Pa})(29 \text{ g/mol})}{(8.314 \frac{\text{Pa m}^3}{\text{g mol K}})(573 \text{ K})} = 791 \frac{\text{g}}{\text{m}^3}$$

$$\dot{m} = \rho A v = \left(791 \frac{\text{g}}{\text{m}^3}\right) \left(\frac{\pi (0.07 \text{ m})^2}{4}\right) (42 \frac{\text{m}}{\text{s}}) \quad A = 3.85 \times 10^{-3} \text{ m}^2$$

$$= 128 \text{ g/s}$$

$$\dot{E}_k = \frac{1}{2} \dot{m} v^2 = \left(\frac{1}{2}\right) \left(128 \frac{\text{g}}{\text{s}}\right) \left(42 \frac{\text{m}}{\text{s}}\right)^2 = 113 \frac{\text{kg} \cdot \text{m}^2}{\text{s}^3}$$

$$= 113 \text{ W}$$

(b) $\rightarrow$ 300°C $\xrightarrow{\quad}$ 400°C

$$\rho = 791 \left(\frac{573}{673}\right) = 674 \frac{\text{g}}{\text{m}^3}$$

$$V_{\text{new}} = ? = \frac{\dot{m}}{\rho A} = \frac{128 \text{ g/s}}{\left(674 \frac{\text{g}}{\text{m}^3}\right) (3.85 \times 10^{-3} \text{ m}^2)} = 49.3 \frac{\text{m}}{\text{s}}$$

$$\Delta\dot{E}_k = \frac{1}{2} (\dot{m}) (V_{400}^2 - V_{300}^2) \quad 42.6 \text{ W}$$

7.6 (7.7 in 4th Ed)

Suppose you pour a gallon of water on a yawling cat 10 ft below your bedroom window.

- How much potential energy (in ft-lbf) does the water have?
- How fast is the water traveling (in ft/s) just before impact?
- True or false: Energy must be conserved, therefore the kinetic energy of the water before impact must equal the kinetic energy of the cat after impact.



$$\cancel{\Delta U} + \Delta E_p + \Delta E_k = \cancel{Q} - \cancel{W}$$

$$\Delta E_p = m g \Delta h$$

$$m = \gamma \rho$$

$$= (1 \text{ gal}) \left(\frac{\text{ft}^3}{7.4805 \text{ gal}} \right) \left(62.43 \frac{\text{lb}_m}{\text{ft}^3} \right)$$

$$= 8.346 \text{ lb}_m$$

$$\Delta E_p = (8.346 \text{ lb}_m) \left(32.2 \frac{\text{ft}}{\text{s}^2} \right) (-10 \text{ ft})$$

$$32.2 \frac{\text{lb}_m \cdot \text{ft}}{\text{lb}_f \text{ s}^2}$$

$$= -83.46 \text{ ft} \cdot \text{lb}_f = -\Delta E_k$$

$$\Delta E_k = \frac{1}{2} m (\cancel{v_2^2} - \cancel{v_1^2}) \rightarrow 0$$