## **Happy Monday!!!**

Fun Video

# **Fatherly Advice**

- · Don't get behind!
- E COLL
- Draw pictures of process

  Try pot to take abortants
  - Try not to take shortcuts
  - Work efficiently
- We will not be using E-Z Solve
- We have Mathcad and Excel
- There is a lot of reading in the workbook on Problem 4.11 (4.17 in 4<sup>th</sup> Ed.), but it is worth it!!
- Workbook pages are optional on all starred problems
- The author throws in some "think about it" problems
  - This coincides with a college initiative on innovation
  - Have fun with it; use engineering intuition



## **Outline for Class 6**

- · Define "Independent Equations"
- Degree of Freedom Analysis (DOF)
  - Procedure
  - Examples
- · Define "Other Relations"
- Please write in the front cover of  $3^{rd}$  Edition:  $\rho_{H2O}$  = 1 g/cm³ = 1000 kg/m³ = 62.4 lb<sub>m</sub>/ft³ = 1 kg/liter



## **Degrees of Freedom Analysis**

- · Method to attack problems
- Kind of like # of eqns = # of unknowns
- · Can tell where to start a problem
- DOF required for all remaining problems in Chapter 4

From my Mother-In-Law:

If you don't listen you gotta feel

## **Independent Equations**



## Who Can Solve the Following Equations?

$$x + 2y = 4$$
$$2x + 4y = 8$$

## **DOF in Chem Eng**

## **Non-Reacting Systems**

- Equations come from material balances
- # independent balance eqns = # species

#### Common mistake:

- Write all species balances <u>plus</u> overall balance
  - Not all independent

# **Example: 2 species**

### Species Balances:

$$x_{A1}m_1 + x_{A2}m_2 = x_{A3}m_3$$
  
 $x_{B1}m_1 + x_{B2}m_2 = x_{B3}m_3$ 

#### **Total Mass Balance**

$$m_1 + m_2 = m_3$$

$$x_{A1} + x_{B1} = 1$$
, etc.



# of Unknowns

# of Equations

## **Additional Equations**

- Equations other than material balance equations are sometimes given
- · Often necessary to solve the problem
- · Relate some of the unknown variables

#### **Example:**

• 95% of the feed ends up in stream 1

$$\dot{m}_1 = 0.95 \dot{m}_{feed}$$

For DOF analysis, DO NOT SOLVE YET!

# DOF for Non-Reacting Systems

## # unknowns

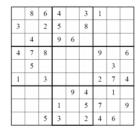
- # independent balance equations
- # of additional relationships

DOF

Formal method
Useful for complicated systems (tells you where to start)

If DOF = 0 good! (unique solution possible)
DOF > 0 No unique solution (too many unknowns)
DOF < 0 Over-specified (too many eqns, or one eqn may not be independent)

## **Analogy to Sudoku**



- How would you set up a logic diagram to solve any sudoku puzzle?
  - Is it solvable?
  - Where do you start?

# Example 1 (on handout)

## **Helpful Hints on DOF**

- 1. Use species flow rates (m<sub>1A</sub>, m<sub>1B</sub>, etc.) if possible instead of mole fractions (y<sub>1A</sub>, etc.)
- 2. Remember that one species mass or mole fraction is not independent ( $\Sigma y_i = 1$ )
- If only one species mass or mole fraction in a stream is unknown, calculate it and treat it as known
- 4. It is often easiest to use the total mass balance as one equation instead of all of the species balances
- Use the flow rate given in the problem as the basisChoose a basis if only mass fractions are given
- 6. If you know the densities and volumetric flow rates, calculate mass flow rates immediately

# Example 2

(on back of handout)



## **Other Examples**

- Prob 4.24 (4.15 in 3<sup>rd</sup> Ed.)
- Prob 4.18 (4.12 in 3<sup>rd</sup> Ed.) (if time)

