

## Class 8

### Simple Coal Devolatilization Models

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## Outline

- 1-step
- 2-step
- DAEM
- Compare capabilities
- Blowing Factor
- Bateman video
- Discuss effects of blowing

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## 1-Step Model

$$\frac{dV}{dt} = k(V_{\infty} - V)$$

- $V$  = % of coal that becomes volatiles
- $V_{\infty}$  = “Ultimate” yield (yield at infinite time)
- $k$  = Arrhenius rate constant  $\{A \exp(-E/RT)\}$

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## Question 1

1. One step devolatilization models are of the form:

$$\frac{dV}{dt} = k(V_{\infty} - V)$$

where  $V$  is the fractional yield of volatiles (mass of volatiles per mass of daf coal), and  $V_{\infty}$  is the hypothetical ultimate yield. Assuming a  $V_{\infty}$  of 0.5, please integrate this expression at  $10^4$  K/s from 300 K to 1400 K for the following two rates:

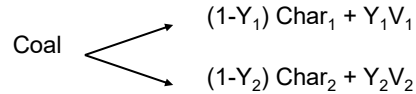
(a) Badzioch & Hawksley (1970):  $k = 3.12 \times 10^5 \exp[-8961/T]$

(b) Solomon, et al. (1976):  $k = 4.3 \times 10^{14} \exp[-27,544/T]$

where  $T$  is in Kelvin and  $k$  is in seconds<sup>-1</sup>. Plot  $V$  versus  $T$ . What is the time and corresponding temperature when the yield reaches  $V = 0.4$ ? You will probably have to integrate this numerically.

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## 2-Competing Step Model



$$\frac{d\text{Coal}}{dt} = -(k_1 + k_2)\text{Coal}$$

$$\frac{dV}{dt} = \frac{dV_1}{dt} + \frac{dV_2}{dt} = (Y_1 k_1 + Y_2 k_2)\text{Coal}$$

- Advantage: Heating rate effect on both rate and yield

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## Ubhayaker Coefficients (buried in a figure)

coal process may contribute to the reaction, with consequent rupturing of unsaturated bonds.  $X_{O_2} = 0$ .

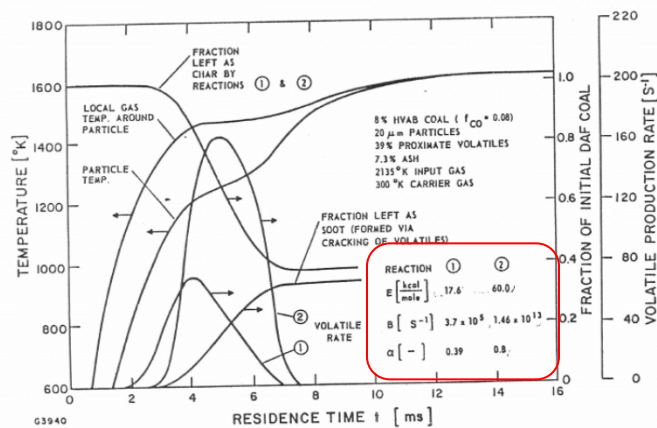


FIG. 2. Numerical solution to the analytical model evaluated for a monodisperse pulverized coal sample. The kinetic parameters were obtained by curve fitting the data of Kimber and Gray<sup>3</sup> and Badzioch and Hawksley.<sup>4</sup>

From Ubhayakar, et al., 16<sup>th</sup> Symp. (Int.) on Comb.  
 p. 480 (1976).

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## Question 2

2. Repeat problem 1 for the 2-step competing model using the following rate constants:

	A <sub>1</sub> sec <sup>-1</sup>	A <sub>2</sub> sec <sup>-1</sup>	E <sub>1</sub> (kcal/mol)	E <sub>2</sub> (kcal/mol)	α <sub>1</sub>	α <sub>2</sub>
Kobayashi 2-step	2.0e5	1.3e7	25	40	0.3	1.0
Ubhayakar 2-step	3.7e5	1.46e13	17.6	60	0.39	0.80

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## Distributed Activation Energy Model (DAEM)

$$\frac{V_{\infty} - V}{V_{\infty}} = \int_0^{\infty} e^{-\int_0^t k dt} F(E) dE$$

$$F(E) = \frac{1}{\sigma\sqrt{2\pi}} e^{-\frac{(E-E_0)^2}{2\sigma^2}}$$

- Assumption: Volatiles can be released from bins of different activation energy **in parallel**
- Advantage: Heating rate effects on rate
- Derivative form available

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## How do you solve the DAEM?

Gaussian quadrature!

$$\int_{-1}^1 f(x) dx = \sum w_i f(x_i)$$

- Break up activation energies into 5 to 10 bins
- Quadrature theory tells what the weighting functions are
- Like 5 to 10 parallel reactions weighted appropriately

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## Series Distributed Activation Energy

- Concept of an effective E
- $E_{\text{eff}}$  changes according to the distribution function:

$$E_{\text{eff}} = f(V, V_{\infty}, \sigma)$$

- $E_{\text{eff}}$  changes according to a Gaussian distribution based on extent of conversion

$$\frac{dV}{dt} = k_{\text{eff}}(V_{\infty} - V)$$

$$k_{\text{eff}} = A e^{-\frac{E_{\text{eff}}}{RT}}$$

- MUCH faster with great results

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### Question 3

3. Why was the distributed activation energy model (DAEM) developed (i.e., what was the underlying idea)? Please find the typos in Equation 3.7 in Smoot and Smith. Repeat Problem 1 for the DAEM model using the following coefficients:
- $$V_{\infty} = 0.5$$
- $$k_0 = 1.67 \times 10^{13} \text{ sec}^{-1}$$
- $$E_0 = 50.65 \text{ kcal/mol}$$
- $$\sigma = 7.01 \text{ kcal/mol}$$

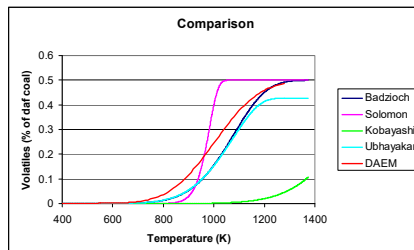
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### Question 4

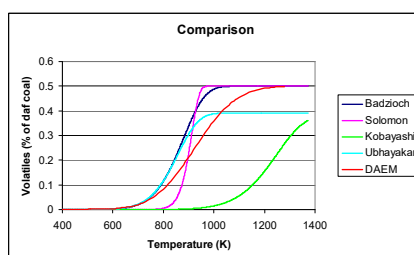
4. Please compare the capabilities of the 3 simple models (1-step, 2-step, and DAEM) with respect to calculation of total volatiles yields as a function of time, temperature, heating rate, pressure, and coal type. You may want to look at the paper by Richards et al., *Fuel*, **185**, 171-180 (2016).

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## Comparison of Coal Devolatilization Models



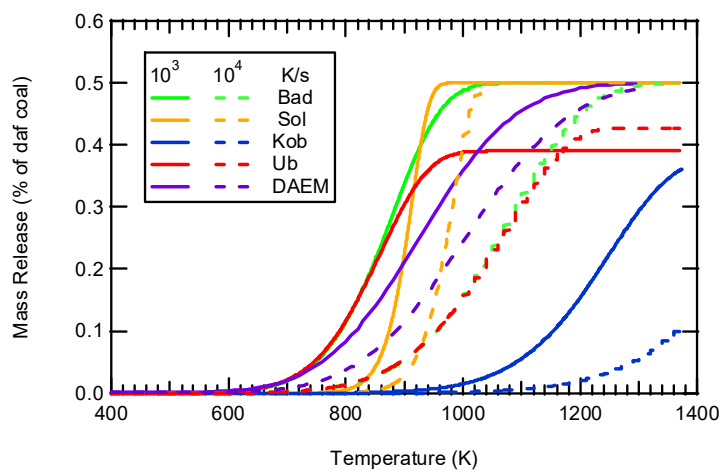
( $10^4$  K/s)



( $10^3$  K/s)

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## Effect of Heating Rate



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## Model Comparison

	# of constants	Yield = $f(\text{heating rate})$ ?	Pressure effects?	T effect?	Coal type?
1-Step	A,E,V*			X	
2-step	A1,E1,Y1 A2,E2,Y2	x		X	
DAEM	A,E0,Sigma a,V*			X	

Industrial practice: Use 2-step,  
 $Y_1$  = ASTM volatiles yield,  
 $Y_2 = 2Y_1$

Result: works OK but not great

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Coal Video

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## Question 5

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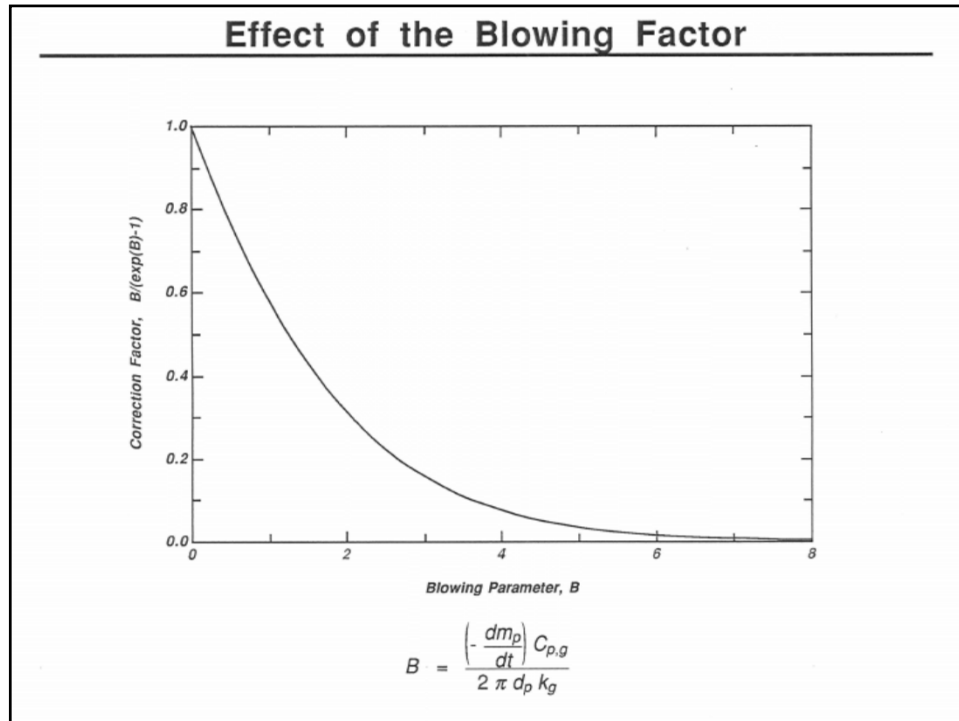
5. Please discuss what the blowing factor is in relation to coal devolatilization. You may want to look at the paper by Fletcher, *Combustion Science and Technology* **63**, 89 (1989).

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## Blowing Factor

See MS Word Handout

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## Question 6

6. If the coal ignites heterogeneously before devolatilization, why will the heterogeneous reaction quickly quit and then restart at a later time?

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