Lecture 20 - NPT Ensemble and Fluctuations

A. The NPT Ensemble

The last ensemble furt we commonly encounter has a fixed value of N,P, i.T. The new thing here is that volume can fluctuate. This is called the isothermal - isobaric ensemble. I will not derive the probability density, but I will give it.

$$\Delta_{NPT} = \frac{1}{3^{2}N^{1}} \int e^{-\beta(H+PV)} dx$$
 (Partition function)
$$\int_{NPT} e^{-\beta[H(x)+PV]} dx$$
 (PDF)
$$\int_{P} e^{-\beta[H(x)+PV]} dx$$

B. Connections Between Ensembles

Mormalized Prob. Density

Partition Function

Thermo Connection

micro canonical ensemble: fixed N, V, E

$$\beta(N,V,E) = \frac{S(H-E)}{\Omega} \qquad \Omega = \frac{1}{1-1} \int S(H-E) dx \qquad S= L_0 \ln \Omega$$

canonical ensemble: fixed N, V, T

$$g(N,V,T) = \frac{e^{-\beta H}}{Q}$$
 $Q = \int_{N}^{\infty} \int_{N}^{\infty} e^{-\beta H} dx$
 $A = -k_0 T \ln Q$

grand canonical ensemble: fixed μ, V, T

Isobanic-Isomermal ensemble: fixed N, P, T

$$\tilde{g}(N,P,T) = \frac{e^{-\beta(H+PV)}}{\Delta}$$

$$\Delta = \frac{1}{h^{N}!} \int e^{-\beta(H+PV)} dx \qquad G = -k_BT \ln \Delta$$

An interesting note: The portition functions can be obtained by Laplace transforms and bu connection formulas by legendre transforms (see Appendix). Laplace transforms and legendre transforms switch between functions with identical information. So, all ensembles have equivalent information.

C. Fluctuations

We have looked af ensemble averages, but what about fluctuations?

For example, in the NVT (canonical ensemble) we have N, v, and T that are fixed, but u, P, and E can fluctuate. We already know the whole PDF, so we should be able to figure out what the variance of these quantities are.

Example: NVT σ_{E}^2 , variance of energy

$$\sigma_{E}^{2} = \langle (H - \langle H \rangle)^{2} = \int \beta (H - \langle H \rangle)^{2} d\underline{x} \qquad \beta = \underbrace{e^{-\beta H}}_{Q}$$

$$= \int \underbrace{e^{-\beta H}}_{Q} (H^{2} - 2H \langle H \rangle + \langle H \rangle^{2}) d\underline{x} \qquad \lambda = \underbrace{h^{2} e^{-\beta H}}_{Q} d\underline{x} \qquad$$

$$= \frac{1}{Q} \int_{A}^{A} \frac{1}{e^{-\beta H}} \int_{A}^{A}$$

$$= -\frac{\int}{Q} \frac{\partial}{\partial \beta} \int H e^{-\beta H} d\vec{x} - \langle H \rangle^2 \qquad \text{T is constant.}$$

$$can pull out \triangleq$$

$$\partial \beta$$

$$= -\frac{1}{Q} \frac{\partial}{\partial \beta} \left(\langle H \rangle Q \right) - \langle H \rangle^2 = -\frac{Q}{Q} \frac{\partial \langle H \rangle}{\partial \beta} - \frac{\langle H \rangle}{Q} \frac{\partial Q}{\partial \beta} - \langle H \rangle^2$$

Simplifying () and (2)

$$0 \frac{\partial \langle H \rangle}{\partial B} = \frac{\partial \langle H \rangle}{\partial T} \frac{\partial T}{\partial B} = \frac{\partial \langle H \rangle}{\partial T} \left(-k_B T^2 \right) = -k_B T^2 \frac{\partial U}{\partial T} = -k_B T^2 C_V$$

Plug into 02:

D. The Thermodynamic Limit

We are interested in the magnitude of these fluotuations as N >00. As we have said many times N=1023 in many cases. The limit N -> 00, V -> 00, N/U = constant is called the "thermodynamic limit" (TI).

What does our distribution look like in the T.L for the NUT ensemble?

We can find out with a cumulant expansion in the limit N>00.

Skipping the details (see appendix), the cumulant expansion of Paux is

$$\beta_{NVT}$$
 = $\frac{e^{-\beta H}}{Q}$ $Q = \frac{1}{2}N_{N!}\int e^{-\beta H}dx = \int \Omega(E)e^{-\beta E}dE$, $H(\underline{x}) = E$

$$K(s) = ln \langle e^{sE} \rangle = k_1 s + \frac{k_2}{2!} s^2 + \frac{k_3}{3!} s^3 + ...$$

$$k_1 = \langle E \rangle = U = -\frac{\partial \ln Q}{\partial \beta}$$

Went from a distribution with 10²³ variables to one with only a few: U, N, T.

$$k_2 = \langle E^2 \rangle - \langle E \rangle^2 = k_B T^2 C_V = \frac{\partial^2 k_B}{\partial \beta^2}$$

Normal Distribution

mean ~ U , var ~ Cv

Example: Ideal gas

K(s) = Us + EBT CV 52 + ...

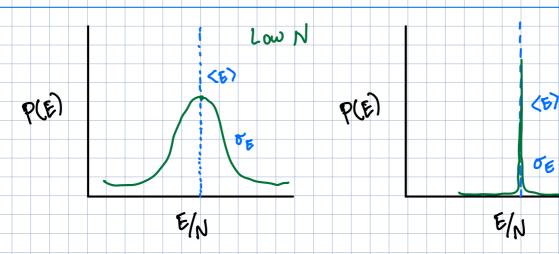
$$\langle E \rangle = \frac{3}{2}Nk_{B}T$$
 $\sigma_{E}^{2} = k_{B}T^{2}C_{V} = k_{B}T^{2}(\frac{3}{2}Nk_{B}) = \frac{3}{2}Nk_{B}T^{2}$

$$\frac{\delta_E}{E} = \frac{\sqrt{\gamma_2}N}{2} \frac{\xi_0 T}{\xi_0 T} = \left(\frac{3}{2}N\right)^{-\frac{1}{2}}$$

This same scaling holds for all cases (not just ideal gas). Internal energy and heat capacity are extensive quantities ($U \sim V$, $C_V \sim N$).

Main takeaway: At large N, ensemble overages grow proportional to N, but fluctuations only grow like N^{r_2} . Thus, the relative size of the fluctuations $\sigma_{N/N} \propto N^{-r_2}$. When $N = co^{23}$, these fluctuations are really small.

High N



Notes:

• the fluctuations don't go to zero, but the relative value of the fluctuations becomes negligible.

Example: lim of #0 lim of 0

- · Fluctuations are still important at small scales (small N, V).
- In canonical N, V, it are fixed and M, P, and E have distributions.

 The fact that the distribution becomes a dirac ofthe for E (or mor P)

 shows that all ensembles are equivalent in the thermodynamic limit.
- It is called the thermodynamic limit because the fluctuations in the system variables: P, V, V, T, M, N become negligible. If it were not 50, then we could not define state variables in thermodynamics. Or, alternatively, our different easembles would not give consistent answers. ("coarse grain" from microscopic D.O.F. to macroscopic state functions)
- The thermodynamic limit is a manifestation of the central limit Theorem for statistical mechanics.

E- Principle of Entopy Maximization

In thermodynamics, we learn that systems tend to maximize their entropy and minimize their free energy (Helmholtz or Gibbs). We

also learn that systems move spontaneously from lower entropy to higher entropy and higher free energy to lower free energy. What can statistical themodynamics teach us about these concepts? It teaches us that probability drives entropy maximization.

Consider an ideal gas in an NVE ensemble where N nolecules are initially prepared in a volume of Vi=1cm3. We open this container in a volume of 1/2= 1m3. The probability of a single microstate having all of the molecules located in the original volume, U = 1cm3 is the exact same as a single microstate where the molecules are spread Annoughout V2= 1m3. This might seem counter intuitive, but it is a consequence of our assumption of a priori equal probabilities. However, there are many more nicrostates where the molecules are spread out! So, if each one is egrally likely, the system will move to the most frequently encountered microstates.

This can still be subject to fluctuations at low N. If N<10, then it is quite possible for all of them to suddenly appear in Vi again. But for N>00, it becomes virtually impossible. Consider the plot I've made (see python plot) for the number of microstates Q(v) relative to Q(Im") for different values of N.

Example: Ideal gas entropy in NVE ensemble

S =
$$lu\left(\frac{V}{N}\frac{1}{\lambda_{10}^{2}}\right)^{\frac{1}{2}} = \frac{V^{N}\left(2\pi m G\right)^{2N/2}}{V^{N}\left(2\pi m G\right)^{2N/2}}$$

$$\frac{h}{2\pi m k_{0}T}, E = \frac{3}{2}Nk_{0}T \rightarrow \lambda_{10} = h\left(\frac{3N}{4\pi m E}\right)^{\frac{1}{2}}$$

Stirlings approximation to get
$$\Omega$$
: (skip math in class)

15 approximation to get
$$\Omega$$
: (Skip math in class)
$$\Gamma\left(\frac{3N}{2}\right) = \left[2\pi \cdot \frac{3N}{2}\right]^{\frac{1}{2}} \left(\frac{3N/2}{e}\right) \qquad N! = \left(2\pi N\right)^{\frac{N}{2}} \left(\frac{N}{e}\right)^{\frac{N}{2}}$$

$$Q = \frac{V^{N}}{h^{3N}} \frac{1}{(2\pi N)^{1/2}} \frac{e^{N}}{N^{N}} (2\pi m E)^{3N/2} \frac{e^{3N/2}}{(2\pi \frac{2N}{2})^{1/2}} \frac{e^{3N/2}}{(3N/2)^{3N/2}}$$

$$= (\frac{1}{N})^{N} \left[\frac{2\pi m E}{2\pi h^{2}} \right]^{\frac{3N}{2}} \left(2\pi \cdot N \cdot 2\pi \cdot \frac{3N}{2} \right)^{\frac{1}{2}}$$

$$= (\frac{1}{N})^{N} \left[\frac{2\pi m E}{2} \cdot h^{2} \right]^{\frac{3N}{2}} \left(2\pi \cdot N \cdot 2\pi \cdot \frac{3N}{2} \right)^{\frac{1}{2}}$$

$$= \left(\frac{V}{N}\right)^{N} \left(\frac{y}{3} \frac{\pi mE}{Nh^{2}}\right)^{3N/2} = \left(\frac{V}{N\lambda_{1}^{3}}\right)^{N} \cdot \frac{e^{5N/2}}{2\pi N} \left(\frac{y}{2}\right)^{2} = \left(\frac{V}{N\lambda_{1}^{3}}\right)^{N} \cdot \frac{e^{5N/2}}{\sqrt{6}\pi N}$$

$$\Omega = \begin{pmatrix} V & N & 6N/2 \\ W & 2^3 \end{pmatrix} \cdot \frac{e}{16 \pi N}$$
 correction

$$N=2$$
 $\Omega(0.9 m^2)/\Omega(1m^3) = 0.81 81:100$

$$N=10$$
 $\Omega(0.9 \text{ m}^3)/\Omega(1\text{m}^3)=0.35$ 35: 100

$$N = (0^2 \Omega(0.9 m^3)/\Omega(1m^2) = 2.6 x(0^5 3:100,000)$$

$$N = 10^3$$
 Q (09m³)/ Ω (1m³) = 2x10 46 2: 10 46

For large N, there is almost no chance the system will spontaneously confine itself to a smaller volume. This is why entropy is maximized. the system will always find itself in the most frequently occuring set of microstates!

F. Appendix: Laplace and Legendre Transforms

Laplace Transform

Legendre Transform

Micro annonical -> canonical

taplace vaniable: E → U

canonical -> grand canonical

$$= = \sum_{n=0}^{\infty} Q 2^{n} - pv = A - \mu N = 2 : fugacity$$

a discrete laplace transform (Z-transform)

cononical -> isobaric-isothermal

G. Appendix: Cumulant Expansion for NVT ensemble

Calculate the moment generating function for pour (E).

$$M(s) = \langle e^{SE} \rangle = \overline{Q} \int_{0}^{\infty} e^{SE} \Omega(E) e^{-BE} dE$$
 The 'Laplace Transform' version of the ensemble

write the cumulant generating function

$$k(s) = ln M(s) = ln \left[\frac{1}{2} \int \Omega(E) e^{(s-p)E} dE \right]$$

Recall that derivatives of K(s) are related to cumulants

$$\frac{d^n k}{ds^n}\Big|_{s=0} = k_n$$
 < non cumulant

$$K(s) = k_1 s + k_2 \frac{s^2}{2!} + k_3 \frac{s^3}{3!} + \cdots \leftarrow cumulant expansion$$

(i) Expressions for the cumulants in terms of E

$$K_{1} = \frac{d}{d\epsilon} \left\{ \ln \left[\frac{1}{Q} \right] \Omega(E) e^{(g-p)E} dE \right] \right\}_{s=0}$$

$$= \frac{d}{ds} \left[\ln \int_{0}^{\infty} \Omega(e^{(g-p)E}) dE - \ln Q \right]_{s=0}$$

$$= 2 \left[\int \Omega e^{(s-p)E} dE \right]^{-1} d \left[\int \Omega e^{(s-p)E} dE \right]$$

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$$K_{1} = \begin{bmatrix} \int_{0}^{\infty} \Omega E e^{(G-B)E} dE \\ \int_{0}^{\infty} \Omega E e^{(G-B)E} dE \end{bmatrix} = \int_{0}^{\infty} \Omega E e^{-BE} dE$$

$$= \int_{0}^{\infty} \Omega E e^{(G-B)E} dE = \langle E \rangle$$

$$k_{2} = \frac{d^{2}k}{ds^{2}} \Big|_{S=0} = \frac{d}{ds} \left(\frac{dk}{ds}\right)\Big|_{S=0} \text{ From above } \frac{dk}{ds} = 0$$

$$\int_{\Omega}^{\infty} \Omega e^{(s-p)E} dE \Big|_{\Omega}^{\infty} \Omega e^{(s-p)E} dE \Big$$

$$\frac{d}{ds} \int \Omega \mathcal{E} e^{i\mathbf{p}\cdot \mathbf{k}} d\mathbf{E} = \int \Omega \mathcal{E}^2 e^{i\mathbf{p}\cdot \mathbf{k}} d\mathbf{E}$$

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$$\frac{$$

(ii) Expressions for cumulants in terms of Q

$$K(s) = ln \left[\frac{1}{Q} \int_{0}^{\infty} \Omega e^{(s-\beta)E} dE \right] \qquad Q(\beta) = \int_{0}^{\infty} \Omega e^{-\beta E} dE$$

$$Q(\beta) = \int_{0}^{\infty} \Omega e^{-\beta E} dE$$

$$K(s) = \ln \left[\frac{\alpha(\beta-s)}{\alpha(B)} \right] = \ln \alpha(\beta-s) - \ln \alpha(\beta)$$

$$K_1 = \frac{dE}{ds} \left[\frac{d}{s=0} \right] = \frac{d \ln \Omega(\beta)}{ds} \left[\frac{d}{s=0} \right] = \frac{d \ln \Omega(\beta)}{ds}$$

$$= \frac{d \ln \Omega(\beta)}{ds} \left[\frac{d}{s=0} \right] = \frac{d \ln \Omega(\beta)}{ds} = \frac{d \ln \Omega(\beta)}{d$$

$$= \frac{d \ln Q(\beta-5)}{d(\beta-5)}\Big|_{S=0} \frac{d \ln Q(\beta)}{d \beta}$$

$$k_1 = -\frac{\partial mQ}{\partial \beta}$$

$$k_2 = \frac{d^2k}{ds^2}\Big|_{s=0} = \frac{d}{ds}\Big[-\frac{d \ln Q(\beta-s)}{d(\beta-s)}\Big]_{s=0} = \frac{d^2}{d(\beta-s)^2}\Big[\ln Q(\beta-s)\Big]_{s=0}$$

$$k_2 = \frac{J^2 J_n Q}{J_p^2}$$
 This patern holds for all k_n

$$k_n = (1)^n \frac{d^n \ln \alpha}{d \beta^n}$$

(iii) Cumulant Expansion

$$k(s) = \langle E \rangle s f \frac{\sigma_E^2}{2} s^2 f ...$$

Gaussian distribution

$$K(5) = -\frac{d \ln Q}{d g} S + \frac{1}{2} \frac{d^2 \ln Q}{d g^2} S^2 + ... = \frac{\infty}{n} (-1)^n \frac{d^n \ln Q}{d g^n} \frac{S^n}{n!}$$

(iv) Cumulant Expansion in terms of intensive variables

The cumulants are extensive (< N). Rescale s and Q to find the distribution in terms of intensive variables

$$k(5) = -\frac{d \ln Q}{d B} S + \frac{d^2 \ln Q}{d B^2} \frac{S^2}{2} - \frac{d^3 \ln Q}{d B^3} \frac{S^3}{6} + \cdots$$

=
$$-\frac{d}{dp} (pleng) \cdot \frac{3}{pi} + \frac{d^2}{dp^2} (pleng) \frac{3^2}{2N^2} - \frac{d^3}{dp^3} (pleng) \frac{3^3}{6N^3} + ...$$

$$k(8) = -\frac{d \ln q}{d B} + \frac{1}{2N} + \frac{d^2 \ln q}{d B^2} + \frac{1}{6N^2} + \frac{d^3 \ln q}{d B^3} + \frac{3}{8} + \dots$$

As N-> 00, the higher order ferms matter less and less. orescaled a N-1/2.