

Chemical Engineering 512

Nuclear Reactor Transient Modeling

Lecture 8

Point Kinetics Introduction



Spiritual Thought

“There is nothing that has come or will come into your family as important as the sealing blessings. There is nothing more important than honoring the marriage and family covenants you have made or will make in the temples of God.”

-Henry B. Eyring



Objectives

- Learn about kinetics input into RELAP
- Practice Combination of Volumes in Series



What is Point Kinetics?

- Used to compute reactor power behavior in a nuclear reactor
- Point kinetics is space independent and uses core-average fluid conditions, weighting factors, and feedback coefficients
- Nodal kinetics is the multidimensional neutron kinetics option



- Kinetics type
 - Point
 - Nodal
- Feedback type
 - Separabl
 - Table3
 - Table4
 - Table3A
 - Table4A

```

*****
*
*                               Kinetics
*
*****
*
* KineticsType      FeedbackType
30000000 point      separabl
* Decay Power React Nrfac
30000001 gamma 1200e+6 0.0 300.
* ModeratorDensity Reactivity
30000501 6.37250 0.006
30000502 46.8710 0.100
30000503 70.3065 0.250
* Temperature Reactivity
30000601 32.0 0.0
30000602 4500.0 0.0
* VolNum      Incr WeightFact LigTempCoef
30000701 514010000 0 0.07391 0.0
30000702 514020000 0 0.18250 0.0
30000703 514030000 0 0.24359 0.0
* HSNu      Incr WeightFact FuelTempCoef
30000801 5141001 0 0.07391 0.0
30000802 5141002 0 0.18250 0.0
30000803 5141003 0 0.24359 0.0
*

```

- Fission product decay type
 - No-gamma
 - Gamma
 - Gamma-ac
- Total Reactor Power
- Initial reactivity
- Delayed neutron fraction/prompt neutron generation time

```

*****
*
*                               Kinetics
*
*****
*
*      KineticsType      FeedbackType
30000000 point      separabl
*
*      Decay  Power  React  NFrac
300000001 gamma 1200e+6 0.0 300.
*
*      ModeratorDensity      Reactivity
30000501 6.37250 0.006
30000502 46.8710 0.100
30000503 70.3065 0.250
*
*      Temperature      Reactivity
30000601 32.0 0.0
30000602 4500.0 0.0
*
*      VolNum      Incr  WeightFact  LigTempCoef
30000701 514010000 0 0.07391 0.0
30000702 514020000 0 0.18250 0.0
30000703 514030000 0 0.24359 0.0
*
*      HSNu      Incr  WeightFact  FuelTempCoef
30000801 5141001 0 0.07391 0.0
30000802 5141002 0 0.18250 0.0
30000803 5141003 0 0.24359 0.0
*

```



- Moderator density vs reactivity table

```

*****
*
*                               Kinetics
*
*****
*
*      KineticsType      FeedbackType
30000000 point          separabl
*      Decay  Power      React  NFrac
30000001 gamma 1200e+6  0 0    300
*      ModeratorDensity  Reactivity
30000501 6.37250         0.006
30000502 46.8710         0.100
30000503 70.3065         0.250
*      Temperature      Reactivity
30000601 32.0            0.0
30000602 4500.0          0.0
*      VolNum      Incr  WeightFact  LigTempCoef
30000701 514010000  0    0.07391    0.0
30000702 514020000  0    0.18250    0.0
30000703 514030000  0    0.24359    0.0
*      HNum      Incr  WeightFact  FuelTempCoef
30000801 5141001   0    0.07391    0.0
30000802 5141002   0    0.18250    0.0
30000803 5141003   0    0.24359    0.0
*

```

- Fuel temperature vs reactivity table

- Volume weighting factors
 - Increment same as heat structure increment
 - If WeightFact is non-zero, LiqTempCoef = 0
- Heat structure weighting factors
 - If WeightFact is non-zero, FuelTempCoef = 0

```

*****
*
*                                     Kinetics
*
*****
*
*      KineticsType      FeedbackType
30000000 point      separabl
*      Decay  Power      React  NFrac
30000001 gamma  1200e+6  0.0    300.
*      ModeratorDensity      Reactivity
30000501 6.37250      0.006
30000502 46.8710      0.100
30000503 70.3065      0.250
*      Temperature      Reactivity
30000601 32.0      0.0
30000602 4500.0      0.0
*      VolNum      Incr  WeightFact  LiqTempCoef
30000701 514010000  0    0.07391    0.0
30000702 514020000  0    0.18250    0.0
30000703 514030000  0    0.24359    0.0
*      HSNu      Incr  WeightFact  FuelTempCoef
30000801 5141001  0    0.07391    0.0
30000802 5141002  0    0.18250    0.0
30000803 5141003  0    0.24359    0.0

```

- Card
1CCCG701
– Source type
 - If 1000-1004, then point kinetics is used

```

*****
*
*                               Heat Structures
*
*****
*
*      AxialHS  RadMesh  GeoType  SSFlag  LeftBound  Reflood
11000000      6        8        2        1        5.74        0
*      MeshLocation      MeshFormat
11000100      0        1
*      NumOfIntervals      RightCoordinate
11000101      7        6.5
*      CompositionNum      IntervalNum
11000201      5        7
*      SourceValue      IntervalNum
11000301      0.0        7
*      InitialTemp      MeshPointNum
11000401      500.        8
*      BoundaryVol/Table  Incr  BCType  SACode  SA/Factor  HSNum
11000501      150010000      0        1        1        10.0        6
*      BoundaryVol/Table  Incr  BCType  SACode  SA/Factor  HSNum
11000601      0        0        1        10.0        6
*      SourceType  Pf      LeftBoundMult  RightBoundMult  HSNum
11000701      1000      0.001791      0.0      0.0      6
*      WordFormat
11000800      0
*      HydDiam  HLFor  HLRev  GSLFor  GSLRev  GLCFor  GLCRev  Boil  HSNum
11000801      0.0      3.0      3.0      0.0      0.0      0.0      0.0      1.0      6
*      WordFormat
11000900      0
*      HydDiam  HLFor  HLRev  GSLFor  GSLRev  GLCFor  GLCRev  Boil  HSNum
11000901      0.0      3.0      3.0      0.0      0.0      0.0      0.0      1.0      6
*

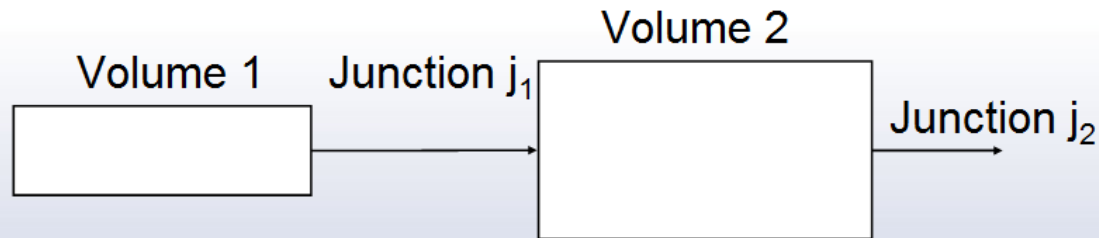
```

Combining Volumes

- Reduce the number of control volumes
- Avoid having control volumes that are too small
- Can combine in series or parallel



Combining in Series



Hydrodynamic:

Length:

Dx_1

Dx_2

Elevation change:

Dz_1

Dz_2

Hydraulic diameter:

D_{h1}

D_{h2}

Area:

A_1

A_{j1}

A_2

A_{j2}

Volume:

V_1

V_2

Form losses:

K_{j1}

K_{j2}

Heat structures:

Inner radius:

R_{i1}

R_{i2}

Thickness:

t_1

t_2

Length:

L_1

L_2

Combining Hydrodynamic

$Dx_m = Dx_1 + Dx_2$	



Combining Hydrodynamic

$Dx_m = Dx_1 + Dx_2$	$Dx_m = 2.0 + 2.0 = 4.0$
$Dz_m = Dz_1 + Dz_2$	



Combining Hydrodynamic

$Dx_m = Dx_1 + Dx_2$	$Dx_m = 2.0 + 2.0 = 4.0$
$Dz_m = Dz_1 + Dz_2$	$Dz_m = 0.0 + 0.0 = 0.0$
$V_m = V_1 + V_2$	



Combining Hydrodynamic

$Dx_m = Dx_1 + Dx_2$	$Dx_m = 2.0 + 2.0 = 4.0$
$Dz_m = Dz_1 + Dz_2$	$Dz_m = 0.0 + 0.0 = 0.0$
$V_m = V_1 + V_2$	$V_m = 0.00789*2+0.01863*2 = 0.05304$
$A_m = V_m/Dx_m$	



Combining Hydrodynamic

$$Dx_m = Dx_1 + Dx_2$$

$$Dx_m = 2.0 + 2.0 = 4.0$$

$$Dz_m = Dz_1 + Dz_2$$

$$Dz_m = 0.0 + 0.0 = 0.0$$

$$V_m = V_1 + V_2$$

$$V_m = 0.00789*2 + 0.01863*2 = 0.05304$$

$$A_m = V_m / Dx_m$$

$$A_m = 0.05304 / 4.0 = 0.01326$$

$$D_{hm} = \left[\frac{\left(\frac{Dx_m}{A_m} \right)^{1.75}}{\frac{Dx_1}{D_{h1}^{1.25} A_1^{1.75}} + \frac{Dx_2}{D_{h2}^{1.25} A_2^{1.75}}} \right]^{0.80}$$



Combining Hydrodynamic

$$Dx_m = Dx_1 + Dx_2$$

$$Dx_m = 2.0 + 2.0 = 4.0$$

$$Dz_m = Dz_1 + Dz_2$$

$$Dz_m = 0.0 + 0.0 = 0.0$$

$$V_m = V_1 + V_2$$

$$V_m = 0.00789 \cdot 2 + 0.01863 \cdot 2 = 0.05304$$

$$A_m = V_m / Dx_m$$

$$A_m = 0.05304 / 4.0 = 0.01326$$

$$D_{hm} = \left[\frac{\left(\frac{Dx_m}{A_m} \right)^{1.75}}{\frac{Dx_1}{D_{h1}^{1.25} A_1^{1.75}} + \frac{Dx_2}{D_{h2}^{1.25} A_2^{1.75}}} \right]^{0.80}$$

$$D_{hm} = \left[\frac{\left(\frac{4}{0.01326} \right)^{1.75}}{\frac{2}{0.102^{1.25} 0.00789^{1.75}} + \frac{2}{0.154^{1.25} 0.01863^{1.75}}} \right]^{0.80} = 0.0777$$

$$K_{jmf} = [(K_{j1}/A_{j1}^2) + (K_{j2}/A_{j2}^2)] A_{jm}^2$$

$$K_{jmr} = [(K_{j1}/A_{j1}^2) + (K_{j2}/A_{j2}^2)] A_{jm}^2$$



Combining Hydrodynamic

$$Dx_m = Dx_1 + Dx_2$$

$$Dx_m = 2.0 + 2.0 = 4.0$$

$$Dz_m = Dz_1 + Dz_2$$

$$Dz_m = 0.0 + 0.0 = 0.0$$

$$V_m = V_1 + V_2$$

$$V_m = 0.00789 \cdot 2 + 0.01863 \cdot 2 = 0.05304$$

$$A_m = V_m / Dx_m$$

$$A_m = 0.05304 / 4.0 = 0.01326$$

$$D_{hm} = \left[\frac{\left(\frac{Dx_m}{A_m} \right)^{1.75}}{\frac{Dx_1}{D_{h1}^{1.25} A_1^{1.75}} + \frac{Dx_2}{D_{h2}^{1.25} A_2^{1.75}}} \right]^{0.80}$$

$$D_{hm} = \left[\frac{\left(\frac{4}{0.01326} \right)^{1.75}}{\frac{2}{0.102^{1.25} 0.00789^{1.75}} + \frac{2}{0.154^{1.25} 0.01863^{1.75}}} \right]^{0.80} = 0.0777$$

$$K_{jmf} = [(K_{j1}/A_{j1}^2) + (K_{j2}/A_{j2}^2)] A_{jm}^2$$

$$K_{jmr} = [(K_{j1}/A_{j1}^2) + (K_{j2}/A_{j2}^2)] A_{jm}^2$$

$$K_{jmf} = [(0.33/0.00789^2)] 0.01863^2 = 1.840$$

$$K_{jmr} = [(0.29/0.00789^2)] 0.01863^2 = 1.617$$



Combining Heat Structures

$$L_m = L_1 + L_2$$

$L_m = L_1 + L_2$	

Combining Heat Structures

$$L_m = L_1 + L_2$$

$$A_{im} = A_{i1} + A_{i2}$$

$$L_m = 2.0 + 2.0 = 4.0$$



Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501*2+0.0770*2)=$$
$$1.597$$

$$R_{im} = A_{im}/(2\pi L_m)$$



Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501*2+0.0770*2)=1.597$$

$$R_{im} = A_{im}/(2\pi L_m)$$

$$R_{im} = 1.597/(2\pi 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$



Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501 \cdot 2 + 0.0770 \cdot 2) = 1.597$$

$$R_{im} = A_{im} / (2\pi L_m)$$

$$R_{im} = 1.597 / (2\pi \cdot 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$

$$V_{hs1} = \pi[(0.0501 + 0.0066)^2 - 0.0501^2]2.0 = 0.00443$$

$$V_{hs2} = \pi[(R_{i2} + t_2)^2 - R_{i2}^2]L_2$$

Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501 \cdot 2 + 0.0770 \cdot 2) = 1.597$$

$$R_{im} = A_{im} / (2\pi L_m)$$

$$R_{im} = 1.597 / (2\pi \cdot 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$

$$V_{hs1} = \pi[(0.0501 + 0.0066)^2 - 0.0501^2]2.0 = 0.00443$$

$$V_{hs2} = \pi[(R_{i2} + t_2)^2 - R_{i2}^2]L_2$$

$$V_{hs2} = \pi[(0.077 + 0.0071)^2 - 0.077^2]2.0 = 0.00719$$

$$V_{hsm} = V_{hs1} + V_{hs2}$$

Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501*2+0.0770*2)=1.597$$

$$R_{im} = A_{im}/(2\pi L_m)$$

$$R_{im} = 1.597/(2\pi 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$

$$V_{hs1} = \pi[(0.0501 + 0.0066)^2 - 0.0501^2]2.0 = 0.00443$$

$$V_{hs2} = \pi[(R_{i2} + t_2)^2 - R_{i2}^2]L_2$$

$$V_{hs2} = \pi[(0.077 + 0.0071)^2 - 0.077^2]2.0 = 0.00719$$

$$V_{hsm} = V_{hs1} + V_{hs2}$$

$$V_{hsm} = 0.00443 + 0.00719 = 0.01162$$

$$R_{om} = \left[R_{im}^2 + \frac{V_{hsm}}{\pi L_m} \right]^{\frac{1}{2}}$$

Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501*2+0.0770*2)=1.597$$

$$R_{im} = A_{im}/(2\pi L_m)$$

$$R_{im} = 1.597/(2\pi 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$

$$V_{hs1} = \pi[(0.0501 + 0.0066)^2 - 0.0501^2]2.0 = 0.00443$$

$$V_{hs2} = \pi[(R_{i2} + t_2)^2 - R_{i2}^2]L_2$$

$$V_{hs2} = \pi[(0.077 + 0.0071)^2 - 0.077^2]2.0 = 0.00719$$

$$V_{hsm} = V_{hs1} + V_{hs2}$$

$$V_{hsm} = 0.00443 + 0.00719 = 0.01162$$

$$R_{om} = \left[R_{im}^2 + \frac{V_{hsm}}{\pi L_m} \right]^{\frac{1}{2}}$$

$$R_{om} = \left[0.0635^2 + \frac{0.01162}{\pi 4.0} \right]^{\frac{1}{2}} = 0.07045$$

$$t = (R_{om} - R_{im})/4$$

Combining Heat Structures

$$L_m = L_1 + L_2$$

$$L_m = 2.0 + 2.0 = 4.0$$

$$A_{im} = A_{i1} + A_{i2}$$

$$A_{im} = 2\pi(0.0501 \cdot 2 + 0.0770 \cdot 2) = 1.597$$

$$R_{im} = A_{im} / (2\pi L_m)$$

$$R_{im} = 1.597 / (2\pi \cdot 4) = 0.0635$$

$$V_{hs1} = \pi[(R_{i1} + t_1)^2 - R_{i1}^2]L_1$$

$$V_{hs1} = \pi[(0.0501 + 0.0066)^2 - 0.0501^2]2.0 = 0.00443$$

$$V_{hs2} = \pi[(R_{i2} + t_2)^2 - R_{i2}^2]L_2$$

$$V_{hs2} = \pi[(0.077 + 0.0071)^2 - 0.077^2]2.0 = 0.00719$$

$$V_{hsm} = V_{hs1} + V_{hs2}$$

$$V_{hsm} = 0.00443 + 0.00719 = 0.01162$$

$$R_{om} = \left[R_{im}^2 + \frac{V_{hsm}}{\pi L_m} \right]^{\frac{1}{2}}$$

$$R_{om} = \left[0.0635^2 + \frac{0.01162}{\pi \cdot 4.0} \right]^{\frac{1}{2}} = 0.07045$$

$$t = (R_{om} - R_{im})/4$$

$$t = (0.07045 - 0.0635)/4 = 0.00174$$

Change Input Deck

- Start with the class example deck.
- Change pipe to represent 3 similar pipes the same values as the original.
- Talk with your group about how you could combine Heat Structures for this example



Combining in Parallel

- DVDs have another example you can work through
- I suggest you make yourself familiar with this process



Assignment

- Watch DVD sections 39-43 before next Class
- Homework 4 is due Thursday (10/2)

